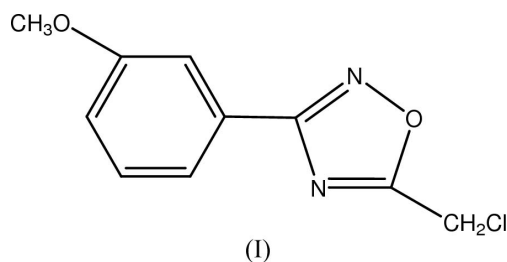


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Key indicators

Single-crystal X-ray study
 $T = 293\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.007\text{ \AA}$
 R factor = 0.069
 wR factor = 0.231
Data-to-parameter ratio = 15.2For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.5-Chloromethyl-3-(3-methoxyphenyl)-
1,2,4-oxadiazoleIn the structure of the title compound, $\text{C}_{10}\text{H}_9\text{ClN}_2\text{O}_2$, there is a
weak intramolecular $\text{C}-\text{H}\cdots\text{N}$ interaction, and inter-
molecular $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{N}$ interactions.Received 14 April 2005
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Comment

1,2,4-Oxadiazoles are an important class of five-membered
heterocycle. Some derivatives of 1,2,4-oxadiazoles have shown
intrinsic analgesic (Terashita *et al.*, 2002), anti-inflammatory
(Nicolaidis *et al.*, 1998) and antipicornaviral (Romero, 2001)
properties. We report here the crystal structure of the title
compound, (I).The molecular structure of (I) is shown in Fig. 1. Selected
bond lengths and angles are given in Table 1. There are weak
intermolecular $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{N}$ interactions (Fig. 2
and Table 2). The combination of these and a weak intra-
molecular $\text{C}-\text{H}\cdots\text{N}$ interaction generates a three-dimen-
sional network.

Experimental

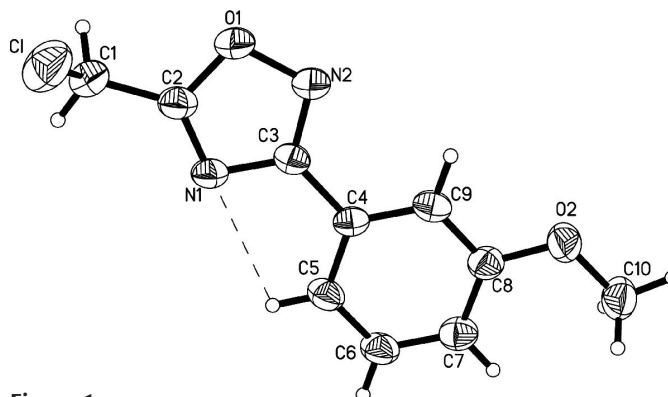
A solution of chloroacetyl chloride (14 mmol) in toluene (10 ml) was
added dropwise to a solution of 3-methoxybenzamidoxime (14 mmol)

Figure 1
A view of the molecular structure of (I), showing displacement ellipsoids
at the 30% probability level. The dashed line indicates the intramolecular
 $\text{C}-\text{H}\cdots\text{N}$ interaction.

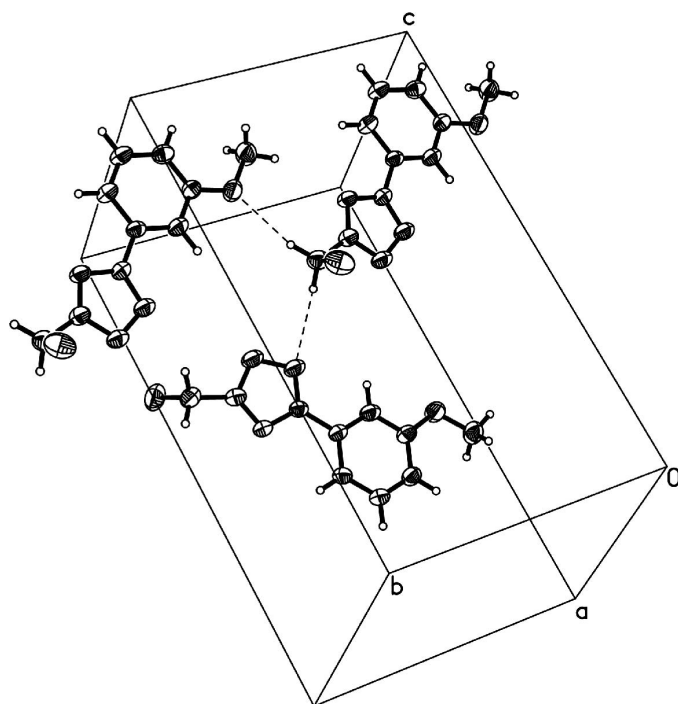


Figure 2
The crystal structure of (I). Dashed lines indicate the intermolecular C—H...O and C—H...N interactions.

in toluene (60 ml). The resulting mixture was refluxed for 6 h and then concentrated under reduced pressure to afford crude compound (I). Pure compound (I) was obtained by recrystallization from a mixture of ethyl acetate (15 ml) and petroleum ether (7.5 ml). Crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of an ethanol solution. $^1\text{H NMR}$ (CDCl_3 , δ , p.p.m.): 7.66–7.68 (*m*, 1H), 7.59–7.60 (*m*, 1H), 7.37–7.40 (*m*, 1H), 7.05–7.07 (*m*, 1H), 4.74 (*s*, 2H), 3.87 (*s*, 3H).

Crystal data

$\text{C}_{10}\text{H}_9\text{ClN}_2\text{O}_2$
 $M_r = 224.64$
Monoclinic, $P2_1/c$
 $a = 7.2350$ (14) Å
 $b = 9.5600$ (19) Å
 $c = 15.454$ (3) Å
 $\beta = 97.03$ (3)°
 $V = 1060.9$ (4) Å³
 $Z = 4$

$D_x = 1.407$ Mg m⁻³
Mo $K\alpha$ radiation
Cell parameters from 25 reflections
 $\theta = 9\text{--}13^\circ$
 $\mu = 0.34$ mm⁻¹
 $T = 293$ (2) K
Block, colourless
 $0.4 \times 0.3 \times 0.3$ mm

Data collection

Enraf–Nonius CAD-4 diffractometer
 $\omega/2\theta$ scans
Absorption correction: none
2234 measured reflections
2066 independent reflections
827 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.036$

$\theta_{\text{max}} = 26.0^\circ$
 $h = 0 \rightarrow 8$
 $k = 0 \rightarrow 11$
 $l = -18 \rightarrow 18$
3 standard reflections every 200 reflections
intensity decay: none

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.069$
 $wR(F^2) = 0.231$
 $S = 1.01$
2066 reflections
136 parameters
H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.087P)^2 + 0.5P]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.21 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.24 \text{ e \AA}^{-3}$

Table 1

Selected geometric parameters (Å, °).

Cl—C1	1.773 (6)	N1—C2	1.285 (6)
O1—C2	1.348 (6)	N1—C3	1.376 (6)
O1—N2	1.407 (6)	N2—C3	1.294 (6)
O2—C8	1.363 (6)	C1—C2	1.465 (8)
O2—C10	1.440 (6)	C3—C4	1.470 (7)
C2—O1—N2	105.8 (4)	N2—C3—N1	114.0 (5)
C8—O2—C10	117.4 (4)	N2—C3—C4	123.3 (5)
C2—N1—C3	103.3 (4)	N1—C3—C4	122.8 (4)
C3—N2—O1	104.0 (4)	C9—C4—C3	120.3 (5)
C2—C1—Cl	109.7 (4)	C5—C4—C3	120.0 (5)
N1—C2—O1	112.9 (5)	O2—C8—C9	116.2 (5)
N1—C2—C1	129.2 (5)	O2—C8—C7	124.6 (5)
O1—C2—C1	117.9 (5)		

Table 2

Hydrogen-bond geometry (Å, °).

$D\text{—}H\cdots A$	$D\text{—}H$	$H\cdots A$	$D\cdots A$	$D\text{—}H\cdots A$
C1—H1B...O2 ⁱ	0.97	2.45	3.393 (4)	166
C1—H1C...N2 ⁱⁱ	0.97	2.54	3.471 (4)	161
C5—H5A...N1	0.93	2.60	2.912 (4)	100

Symmetry codes: (i) $x, y - 1, z$; (ii) $-x, y - \frac{1}{2}, -z + \frac{1}{2}$.

All H atoms were positioned geometrically, with C—H distances in the range 0.93–0.97 Å, and included in the refinement in a riding-model approximation, with $U_{\text{iso}}(\text{H}) = 1.2$ or $1.5U_{\text{eq}}$ of the carrier atom.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *SHELXL97*.

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